## NEW CHIRAL BUILDING BLOCKS BY MICROBIAL ASYMMETRIC REDUCTION: A DIRECT ACCESS TO FUNCTIONALIZED 2R,3R- AND 2S,3R-2-METHYL-3-HYDROXY BUTYRATE SYNTHONS<sup>1</sup>

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Abstract: The reduction of 4-O-benzyl-2-methyl-3-oxo-butyrate esters by several fungal strains produces predominantly (2S,3R)-<u>anti</u>-hydroxyesters in high optical purity. The corresponding (2R,3R)-<u>syn</u> isomer was obtained, with other strains, only in mixture with the <u>anti</u> isomer. These esters are useful precursors for chiral 2-methyl-3-hydroxy-butyrolactone synthons.

Functionalized 2-methyl-3-hydroxy butyric acids or esters of definite stereochemistry constitute useful building blocks for the synthesis of polyhydroxylated methylated natural products biosynthetically derived from polypropionate-polyacetate pathways. The direct microbial reduction of 2-methyl-3-oxosuccinate esters **1a** by various yeast strains has been described to give generally (2R,3R)- and (2S,3R)-isomer mixtures, which are difficult to separate, in variable optical purity<sup>2,3</sup>. Several attempts have been made to differentiate the two ester functions and to replace the 4-carboalkoxy function by a group considered as a carboxylic group (or other functional group) equivalent. Benzylidene (**1b**)<sup>4</sup>, furyl (**1c**)<sup>5</sup> and thienyl (**1d**)<sup>6</sup> substituted compounds have been tested as microbial reduction substrates, with moderate success : in most cases, notwithstanding the easy enolization of such esters, mixtures of diastereoisomers were obtained, indicating a low selectivity towards the enantiomers of the keto-substrate. Moreover , despite the easier separation of such isomers, the ozonolysis procedure required for the conversion to a carboxylic acid rendered this method difficult in practice<sup>3</sup>. One attempt to use enantioselective hydrolysis of the corresponding racemic 3-acetoxy esters by lipases has been recently described<sup>7</sup>.

$$\frac{1a}{R} : R = COOMe \text{ or } COOEt; R' = Me \text{ or } Et$$

$$\frac{1b}{R} : R = CH=CH=C6H5; R' = Et$$

$$\frac{1c}{R} : R = \swarrow R' = Me$$

$$\frac{1d}{R} : R' = \swarrow R' = Me$$

4-O-Benzyl-2-methyl-3-oxoesters such as <u>4a-e</u> were easily obtained by condensation of the acid chloride <u>3</u> with the anion of the methylmalonic acid disilyl ester  $2^8$ , followed by hydrolysis and carbodiimide esterification with various alcohols (Scheme 1).



The reduction of the ethyl ester  $\underline{4b}$  by yeasts or fungi precedently used for the diastereoselective reduction of simpler  $\alpha$ -methyl- $\beta$ -oxoesters<sup>9</sup> gave moderate yields of reduced product, probably because of

		Yield	<u>syn/anti</u>	<u>syn</u>		anti	
	Ester	%	ratio	config.	ee%	config.	ee%
1- <u>Saccharomyces cerevisiae</u> <sup>c</sup>	<u>4a</u>	50	10/90	2R,3R <sup>a</sup>	68	2S,3R <sup>b</sup>	6
	<u>4b</u>	46	54/46	2R,3R	36	2R,3S	24
	<u>4c</u>	64	24/76	2R,3R	96	2R,3S	50
2- Rhodotorula mucilaginosad,e	<u>4b</u>	-	30/70	2R,3R	98	2S,3R	18
	<u>4c</u>	75	43/57	2R,3R	97	2S,3R	86
	<u>4d</u>	20	80/20	-	-	-	-
3- <u>Curvularia lunata</u> d,f	<u>4b</u>	20	5/95	-	-	2S,3R	96
	<u>4c</u>	52	31/69	2R,3R	80	2S,3R	92
	<u>4e</u>	no re	no reduction				
4- <u>Geotrichum candidum</u> d,e	4a	66	2/98	2R,3R	60	2S,3R	96
	<u>4b</u>	70	2/98	-	-	2S,3R	90
	<u>4d</u>	50	4/96	-	-	2S,3R	82
,	<u>4e</u>	no reduction					
5- <u>Mucor racemosus</u> <sup>d,e</sup>	<u>4a</u>	50	40/60	2R,3R	42	2R,3S	6
4 -	<u>4b</u>	40	48/52	2R,3R	90	2R,3S	7
6- <u>Rhizopus</u> <u>arrhizus</u> <sup>d,g</sup>	<u>4b</u>	-	4/96	-	-	2S,3R	81

Table 1: Microbial reduction of 4-O-benzyl-2-methyl-3-oxobutyrate esters 4

a) 2R,3R = 5; b) 2S,3R = 6; c) Baker's yeast (5.5 g) incubated with ester (250 mg) and saccharose (3 g) in water (100 ml) during 2-3 days at 30°C; d) cells or mycelium freshly grown in complex culture medium (100 ml)<sup>9</sup>, then added with ester (100 mg) and incubated at 27°C until complete reduction (2-4 days); e) local strain; f)NRRL 2380; g) ATCC 11145.

some hydrolytic activity of the cells. Moreover the ratio and optical purity of the recovered <u>syn</u> and <u>anti</u> hydroxyesters <u>5b</u> (and/or <u>ent-5b</u>) and <u>6b</u> (and/or <u>ent-6b</u>)<sup>10</sup> were dependent on the microorganism (Table 1). These isomers were easily separated by column chromatography<sup>11</sup> and converted to <u>trans</u> and <u>cis</u> lactones by debenzylation (H<sub>2</sub>/Pd) followed by cyclization (CF<sub>3</sub>COOH/CH<sub>2</sub>Cl<sub>2</sub>) (Scheme 2).



Scheme 2

The configuration of these lactones<sup>12</sup> was determined by comparison with authentic samples of (2S,3S)-lactone **Z** and (2R,3S)-lactone **B**, either directly by GC on a chiral column <sup>13</sup> or after derivatization with MTPA<sup>14</sup> and HPLC separation (Fig.1). These lactones have been obtained in a 98:2 ratio by monoalkylation <sup>15</sup> of the dianion of the lactone **2** (Scheme 3) and separated by medium pressure chromatography<sup>16</sup>.



Scheme 3

Some selected fungal strains are highly enantioselective, producing high amounts of the single (2S,3R)-<u>anti</u>-isomer **6b**, generally in a good optical purity (Table 1, entries 3, 4 and 6); this hydroxyester has been converted to the <u>ent-8</u> lactone. This method is thus complementary to the known alkylation of lactones like **2** <sup>15a</sup> to <u>trans</u> lactones like **7**, which have been used as intermediates in the synthesis of <u>syn</u>  $\alpha$ -substituted  $\beta$ -hydroxyesters<sup>15b</sup>; the microbial method compares advantageously with the alkylation of 3-hydroxyesters, which has been currently used for the preparation of optically active <u>anti</u>  $\alpha$ -substituted  $\beta$ -hydroxyesters<sup>17,18</sup>. However, none of our strains was able to produce a prevailing <u>syn</u>-isomer, nor substantially enriched (3S)-enantiomers.

The effect of a modification of the ester group upon the stereoselectivity of the reduction was investigated (Table 1). In some cases, specially with yeasts, some increase in the enantioselectivity of the carbonyl group reduction was observed with higher esters  $\underline{4c}$ - $\underline{d}$ . On the other hand, when the size of the alkoxy group was increased, the diastereoselectivity of the reduction was significantly shifted to the <u>syn</u> isomer, as precedently reported <sup>19</sup>; however, with <u>R.mucilaginosa</u>, which produced more <u>syn</u> ester, but in low yields (Table 1, entry 2), it appears that this effect may be the result of selective hydrolytic activities rather than a modification of the reduction stereochemistry. Moreover, when the size of the ester group was too large, the reduction was strongly inhibited.

In conclusion, one of the isomeric hydroxyesters (anti-2S,3R- $\underline{6a}^{21}$ , for example with <u>G.candidum</u>) could be obtained in good yield and sufficient optical purity for synthetic applications <sup>15b</sup>. The <u>syn</u>-2R,3R

Figure 1: HPLC separation of MTPA esters of enantiomers of lactones <u>7</u> and <u>8</u>, obtained by cyclisation of mixtures of hydroxyesters

(A) obtained by NaBH4 reduction of ester 4b.

(B) obtained by <u>M. racemosus</u> reduction of ester  $\underline{4b}$ .

Zorbax 5µ silica column (25 cm x 4.6 mm) solvent : isooctane-EtOAc (85/15), 1.5 ml/min



isomer was obtained, in high optical purity, only in mixture with the <u>anti</u> isomer. On the other hand, the (3S) isomers could be obtained, but their enantiomeric purity is always low.

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10- **<u>5h</u>** :  $\delta$  (CHCl<sub>3</sub>) 7.32 (s,5H,ÅrH), 4.55 (s,2H,ArCH<sub>2</sub>), 4.14 (q,J=7Hz,2H,COOCH<sub>2</sub>), 4.05 (m,1H,CHOH), 3.51 (2d,J=4Hz and 6Hz,2H,OCH<sub>2</sub>), 2.65 (dq, J<sub>1</sub>=6Hz,J<sub>2</sub>=5Hz,1H,CHMe), 1.60 (s,1H,OH), 1.21 (t,J=7Hz,3H,CH<sub>3</sub>), 1.18 (d,J=6Hz,3H,CH<sub>3</sub>) ; **<u>6h</u>**:  $\delta$  (CHCl<sub>3</sub>) 7.30 (s,5H,ArH), 4.55 (2d,J=11Hz,2H,Ar-CH<sub>2</sub>), 4.14 (q,J=7Hz,2H,COOCH<sub>2</sub>), 3.85 (ddd, J<sub>1</sub>=7Hz,J<sub>2</sub>=6Hz,J<sub>3</sub>=4Hz,1H, CHOH), 3.57 (dd,J<sub>1</sub>=9Hz,J<sub>2</sub>=4Hz,1H,OCH<sub>2</sub>), 3.48 (dd,J<sub>1</sub>=9Hz,J<sub>2</sub>=6Hz,1H,OCH<sub>2</sub>), 2.75 (dq,J<sub>1</sub>=J<sub>2</sub>=7Hz,1H,CHMe), 1.25 (t,J=7Hz,3H,CH<sub>3</sub>), 1.18 (d,J=7Hz,3H,CH<sub>3</sub>)

11- Silicagel 60H (Merck); solvent : Cyclohexane:AcOEt- 8:2

12-These lactones have been obtained recently in mixtures (see ref.3) by acidic cyclization of the BH<sub>3</sub>-Me<sub>2</sub>S reduction product of diethyl 2-methyl-3-hydroxysuccinates, produced by microbial reduction of the corresponding 3-oxoester, and were similarly used for the identification of the isomeric hydroxyesters.

13- GC of the hydroxylactones for separation of the <u>cis</u> enantiomers or their isopropyl carbamates (see ref.20) for the separation of the <u>trans</u> enantiomers on a Chrompack XE60-polysiloxane S-valine-S- $\alpha$ -phe-nylethylamide silica capillary column (50 m x 0.25 mm).

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21- $[\alpha]_{D}^{21}$  = +24,6° (c=1,3 MeOH) after distillation. ee: 96 %.